

Steroidal Aphidicolin Analogues Derived from Pregnenolone

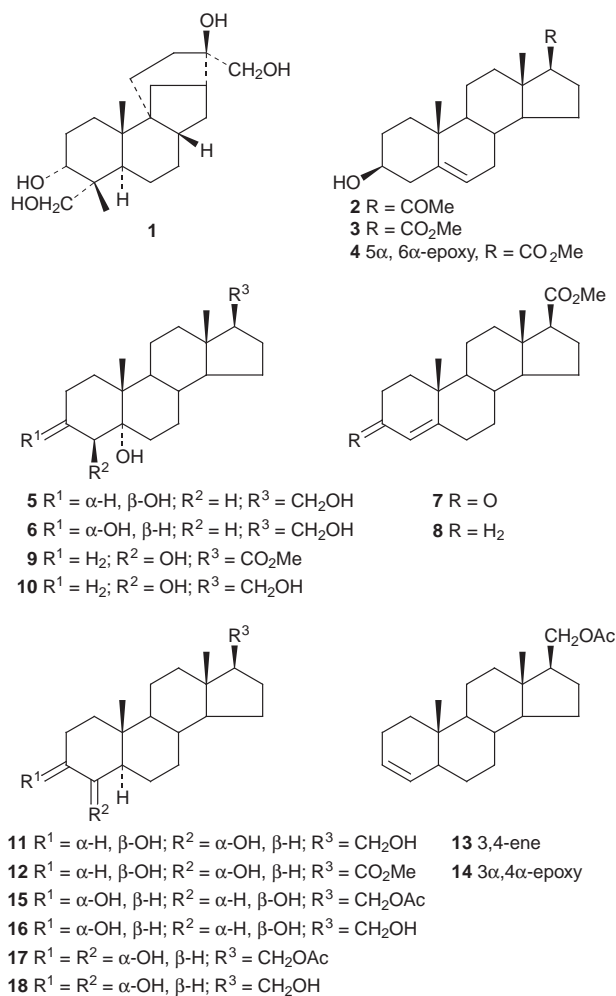
James R. Hanson* and Kudret Yildirim

School of Chemistry, Physics and Environmental Science, University of Sussex, Brighton Sussex, BN1 9QJ, UK

J. Chem. Research (S),
1999, 698–699
J. Chem. Research (M),
1999, 2975–2991

The conversion of pregnenolone into the 3 β ,5 α -, 3 α ,5 α -, 4 β ,5 α -, 3 α ,4 α -, 3 α ,4 β - and 3 α ,4 α -dihydroxy derivatives of 17 β -hydroxymethyl-5 α -androstane as steroidal analogues of the diterpenoid DNA polymerase α inhibitor, aphidicolin, is described.

The steroid hormones exert their biological activity by binding to steroid receptors which also have a nucleic acid binding domain. This binding may result in the initiation of nucleic acid biosynthesis.¹ The diterpenoid aphidicolin **1**² has attracted considerable interest because it is a specific inhibitor of DNA polymerase α .^{3,4} Although aphidicolin is relatively inaccessible, the limited structure–activity studies that have been carried out suggest that the activity may be related to the separation between the hydroxy groups on the A and D rings of aphidicolin.⁵ There is a formal similarity between the rings A, B and C of aphidicolin and the steroid framework and hence we have examined the possibility of mimicking the separation of these hydroxy groups on the more readily available steroids with the object of producing steroid hormone analogues of



aphidicolin.^{6,7} Molecular modelling studies reveal that the superimposition of a steroidal 17-hydroxymethyl substituent onto the 17-hydroxymethyl group of aphidicolin brings steroidal ring A alcohols within the orbit of those of aphidicolin. The preparation of a series of 17-hydroxymethyl steroids from the readily available pregnenolone **2** forms the subject of this paper.

The 17 β -hydroxymethyl group was generated by oxidation of pregnenolone **2** with sodium hypobromite to form the 17-carboxylic acid^{8,9} and methylation with caesium fluoride and methyl iodide in dimethylformamide to generate the methyl ester **3**.^{10,11} Reduction of the 17 β -carbomethoxy group was carried out at appropriate stages in the subsequent sequences.

Epoxidation of 17 β -carbomethoxyandrost-5-en-3 β -ol **3** with *m*-chloroperbenzoic acid gave the 5 α ,6 α -epoxide **4** which was reduced with lithium aluminium hydride to form the triol, 3 β ,5 α -dihydroxy-17 β -hydroxymethylandrostane **5**. Inversion of the 3 β -hydroxy group by the Mitsunobu procedure, was carried out with monochloroacetic acid as the nucleophile.¹¹ Hydrolysis of the bis(monochloroacetate) with aqueous potassium carbonate gave 3 α ,5 α -dihydroxy-17 β -hydroxymethylandrostane.

17 β -Carbomethoxyandrost-5-en-3 β -ol **3** was oxidized to the unsaturated ketone, 17 β -carbomethoxyandrost-4-en-3-one **7**, using the Meerwein–Ponndorf procedure.¹² Reduction with sodium tetrahydroborate in trifluoroacetic acid–acetic acid–acetonitrile¹³ gave 17 β -carbomethoxyandrost-4-ene **8**. Epoxidation of this and hydrolysis of the epoxide with aqueous periodic acid gave the, 4 β ,5 α -diol **9**. Reduction of the 17 β -ester with lithium aluminium hydride gave 4 β ,5 α -dihydroxy-17 β -hydroxymethylandrostane **10**.

Hydroboration of the unsaturated ketone **7** gave 3 β ,4 α -dihydroxy-17 β -hydroxymethyl-5 α -androstane **11** and the corresponding 17 β -carbomethoxy ester **12**. Although reduction of 17 β -carbomethoxyandrost-4-en-3-one **7** with boron trifluoride etherate–sodium borohydride in diglyme followed by treatment with acetic anhydride¹⁵ gave 17 β -acetoxymethyl-5 α -androst-3-ene **13** directly, a better yield of the 3-ene was obtained by carrying out the reaction in a stepwise sequence *via* 3 β -acetoxy-17 β -carbomethoxyandrost-4-ene. Epoxidation of the 3-ene **13** gave the 3 α ,4 α -epoxide **14**. This epoxide was hydrolysed with aqueous periodic acid to the diaxial 3 α ,4 β -diol and then with aqueous potassium carbonate to give 3 α ,4 β -dihydroxy-17 β -hydroxymethyl-5 α -androstane **16**. Catalytic osmylation of the 3-ene using potassium hexacyanoferrate(III) as the co-oxidant¹⁶ gave the 3 α ,4 α -diol **17** from which 3 α ,4 α -dihydroxy-17 β -hydroxymethyl-5 α -androstane **18** was obtained by hydrolysis with aqueous potassium carbonate.

K. Y. thanks Sakarya University, Turkey, for study leave and financial support. We thank Miss L. Abel for carrying out the molecular modelling calculations.

* To receive any correspondence.

Techniques used: IR, ¹H NMR spectroscopy, chromatography

References: 17

Received, 13th July 1999; Accepted, 10th September 1999
Paper E/9/05656D

References cited in this synopsis

- 1 For a review, see: T. Ojasoo, J. P. Raynaud and J. P. Mornon, in *Comprehensive Medicinal Chemistry*, ed. C. Hansch, Pergamon Press, Oxford, 1990, vol. 3 ch. 16.3, p. 1175.
- 2 W. Dalziel, B. Hesp, K. M. Stevenson and J. A. J. Jarvis, *J. Chem. Soc., Perkin Trans. 1*, 1973, 2841.
- 3 S. Ikegami, T. Taguchi and M. Ohashi, *Nature (London)*, 1978, **275**, 458.
- 4 R. Sheaff, D. Ilsley and R. Kuchta, *Biochemistry*, 1991, **34**, 8590.
- 5 S. Hiranuma, T. Shimizu, H. Yoshioka, K. Ono, H. Nakane and T. Takahashi, *Chem. Pharm. Bull.*, 1987, **35**, 1641.
- 6 J. A. Boynton and J. R. Hanson, *J. Chem. Soc., Perkin Trans. 1*, 1995, 2189.
- 7 J. R. Hanson, P. B. Hitchcock and C. Uyanik, *J. Chem. Res.* 1998, (S) 682; (M) 2862.
- 8 J. Staunton and E. J. Eisenbraun, *Org. Synth.*, 1973, **Coll. Vol. V**, 8.
- 9 S. Danishefsky, K. Nagasawa and N. Wang, *J. Org. Chem.*, 1975, **40**, 1989.
- 10 T. Sato, J. Otera and H. Nozaki, *J. Org. Chem.*, 1992, **57**, 2166.
- 11 M. Saiah, M. Bessodes and K. Antonakis, *Tetrahedron Lett.*, 1992, **33**, 4317.
- 12 J. F. Eastham and R. Teranishi, *Org. Synth.*, **Coll. Vol. IV**, 192.
- 13 E. Winterfeldt, U. Tibtamm, R. Hofmeister and H. Laurent, *Ger. Pat.*, DE 3909770, 1990.
- 14 J. R. Hanson, P. B. Hitchcock, M. D. Liman and S. Nagaratnam, *J. Chem. Soc., Perkin Trans. 1*, 1995, 2183.
- 15 R. C. Cambie, P. S. Rutledge, D. W. Scott and P. D. Woodgate, *Aust. J. Chem.*, 1979, **32**, 695.
- 16 M. Minato, K. Yamamoto and J. Tsuji, *J. Org. Chem.*, 1990, **55**, 766.